

Fact Sheet Peroxide-Forming Chemicals

Overview:

Chemicals listed as peroxide---formers have the potential to form explosive crystals under certain circumstances. Stored chemicals that go unused over several months and are exposed to air, heat, moisture, or light can contribute to a potential explosion or injury simply by unscrewing the cap or disturbing the container. It is important to inspect containers for any observable crystals or solids present and to periodically test for non---observable formations of peroxides. **CONTACT ENVIRONMENTAL HEALTH AND SAFETY, OR THE SCIENCE LABORATORY DIRECTOR IF PEROXIDES ARE VISIBLE OR IF THERE IS A POSITIVE RESULT WHEN TESTING FOR PEROXIDES**

Instructions:

- 1. Check to see if your laboratory works with any peroxide---forming chemicals. Most chemical manufacturers will indicate, "may form explosive peroxides" on the label.
- 2. Once the peroxide formers are identified, a label must be placed on each container indicating the "Date Received," "Date Opened," and "Expiration Date." This label will be used to determine when to begin testing and when it is time to dispose of the chemical container.
- 3. The allowable time period for use of the chemical may be extended after peroxide testing. This testing is to be completed by laboratory personnel. Peroxide testing strips are available in the Science Department and labels are available through Environmental Health and Safety.
- 4. Contact Environmental Health and Safety for waste peroxides at x4117 or safety@jjay.cuny.edu.

Test strips are available from the Science Department



Label Peroxide Formers

Date Received	Date Open	Expiration
Date Tested/Result: _		
Date Tested/Result: _		
Date Tested/Result: _		
Date Tested/Result:		

Some Peroxide Forming Chemicals

Group A ChemicalsTest for peroxides before using; discard after 3 months				
Butadiene	Potassium metal			
Chloroprene	Sodium amide			
Divinyl acetylene	Tetrafluoroethylene			
Isopropyl ether	Vinylidene chloride			
Potassium amide				

Group B and C Chemicals Test for peroxides every 6 months; discard after 12 months				
Acetal	2Cyclohexene1ol	Furan	Stryene	
Acetaldehyde	1,4-Dioxane	4Heptanol	Tetrahydrofuran	
Benzyl alcohol	Decahydronapthalene	2Hexanol	Tetrahydronapthalene	
2Butanol dioxanes	Diacetylene	Methyl acetylene	Vinyl acetate	
Chlorobutadiene	Dicyclopentadiene	3Methyl1butanol	Vinyl acetylene	
Chlorofluoroethylene	Diethylene glycol dimethyl ether	Methyl isobutyl ketone	Vinyl chloride	
Cumene	Diethyl ether	4Methyl2pentanol	Vinyl ethers	
Cyclohexene	Ethylene glycol ether acetates	4Pentene1ol	Vinyl pyridine	
Cyclopentene	Ethylene glycol monomethyl ether	2Phenylethanol		